

Table 2. Selected geometric parameters (Å, °)

C(1)—C(2)	1.381 (9)	C(4')—C(5')	1.383 (11)
C(1)—C(6)	1.401 (10)	C(5')—C(6')	1.367 (9)
C(1)—N(7)	1.421 (9)	N(7)—C(8)	1.291 (9)
C(2)—C(3)	1.370 (10)	C(8)—N(9)	1.402 (10)
C(3)—C(4)	1.370 (13)	C(8)—N(10)	1.359 (8)
C(4)—C(5)	1.377 (12)	N(10)—C(11)	1.469 (8)
C(5)—C(6)	1.384 (11)	N(10)—C(15)	1.462 (10)
C(1')—C(2')	1.391 (8)	C(11)—C(12)	1.517 (9)
C(1')—C(6')	1.398 (11)	C(12)—O(13)	1.444 (10)
C(1')—N(9)	1.398 (7)	O(13)—C(14)	1.432 (8)
C(2')—C(3')	1.370 (9)	C(14)—C(15)	1.500 (9)
C(3')—C(4')	1.362 (12)		
C(6)—C(1)—N(7)	121.2 (6)	C(1')—C(6')—C(5')	120.0 (6)
C(2)—C(1)—N(7)	118.6 (6)	C(1)—N(7)—C(8)	120.7 (6)
C(2)—C(1)—C(6)	119.9 (6)	N(7)—C(8)—N(10)	120.1 (6)
C(1)—C(2)—C(3)	119.7 (7)	N(7)—C(8)—N(9)	124.1 (6)
C(2)—C(3)—C(4)	122.0 (8)	N(9)—C(8)—N(10)	115.9 (6)
C(3)—C(4)—C(5)	118.2 (8)	C(1')—N(9)—C(8)	121.8 (5)
C(4)—C(5)—C(6)	122.0 (7)	C(8)—N(10)—C(15)	119.1 (6)
C(1)—C(6)—C(5)	118.4 (7)	C(8)—N(10)—C(11)	122.2 (6)
C(6')—C(1')—N(9)	118.7 (5)	C(11)—N(10)—C(15)	110.5 (5)
C(2')—C(1')—N(9)	123.3 (6)	N(10)—C(11)—C(12)	111.2 (6)
C(2')—C(1')—C(6')	118.0 (6)	C(11)—C(12)—O(13)	110.1 (6)
C(1')—C(2')—C(3')	120.4 (6)	C(12)—O(13)—C(14)	110.6 (5)
C(2')—C(3')—C(4')	121.8 (7)	O(13)—C(14)—C(15)	111.7 (6)
C(3')—C(4')—C(5')	118.0 (7)	N(10)—C(15)—C(14)	110.4 (6)
C(4')—C(5')—C(6')	121.7 (7)		
C(2)—C(1)—N(7)—C(8)	133.0 (7)		
C(6)—C(1)—N(7)—C(8)	-53.1 (9)		
C(1)—N(7)—C(8)—N(9)	-17.2 (10)		
C(1)—N(7)—C(8)—N(10)	161.0 (6)		
N(7)—C(8)—N(9)—C(1')	-69.2 (9)		
N(10)—C(8)—N(9)—C(1')	112.6 (7)		
C(8)—N(9)—C(1')—C(6')	-174.3 (6)		
C(8)—N(9)—C(1')—C(2')	3.8 (9)		
N(7)—C(8)—N(10)—C(11)	145.6 (6)		
N(7)—C(8)—N(10)—C(15)	0.2 (9)		
N(9)—C(8)—N(10)—C(11)	-36.0 (8)		
N(9)—C(8)—N(10)—C(15)	178.5 (6)		

H atoms were refined isotropically, except for those bonded to C2, C5, C11 and C14. These did not refine realistically and were included in the model in their ideal positions (Sheldrick, 1976). The *R* factor is relatively high due to the quality of data collected.

Data collection: *SDP* (Frenz, 1978). Cell refinement: *SDP*. Data reduction: *SDP*. Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985). Program(s) used to refine structure: *SHELX76* (Sheldrick, 1976). Molecular graphics: *ORTEPII* (Johnson, 1976). Software used to prepare material for publication: *PARST* (Nardelli, 1983).

KS thanks the Deutscher Akademischer Austauschdienst for a grant to stay in Berlin, and LS is grateful to CSIR, New Delhi, India, for financial support.

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: KA1123). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

## References

- Frenz, B. A. (1978). *The Enraf-Nonius CAD-4 SDP - a Real-Time System for Concurrent X-ray Data Collection and Crystal Structure Solution*. *Computing in Crystallography*, edited by H. Schenk, R. Olthof-Hazekamp, H. van Koningsveld & G. C. Bassi, pp. 64–71. Delft Univ. Press.

Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.

Maryanoff, C. (1985). *Chem. Week*, (September 16), pp. 13–14.

Nardelli, M. (1983). *Comput. Chem.* **7**, 95–98.

Sheldrick, G. M. (1976). *SHELX76. Program for Crystal Structure Determination*. Univ. of Cambridge, England.

Sheldrick, G. M. (1985). *SHELXS86. Program for the Solution of Crystal Structures*. Univ. of Göttingen, Germany.

Wong-Ng, W., Nyburg, S. C., Awwal, A., Jankie, R. & Kresge, A. J. (1982). *Acta Cryst.* **B38**, 559–564.

*Acta Cryst.* (1995). **C51**, 2331–2333

## Chiral *N*-(6-Amino-3-pyridyl)-*N'*-bicycloalkyl-*N''*-cyanoguanidine Derivative: a Novel Potassium-Channel Opener

HIROSHI SAITO,\* MASAHIRO EDA, SHIGETOSHI SUGIO AND YASUO UEDA

Research Division, The Green Cross Corporation, 2-25-1 Shodai-Ohtani, Hirakata, Osaka 573, Japan

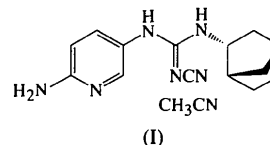
(Received 10 May 1994; accepted 4 January 1995)

## Abstract

The crystal structure of (+)-1-(6-amino-3-pyridyl)-3-[(1*S*\*,2*R*\*,4*R*\*)-bicyclo[2.2.1]hept-2-yl]-2-cyanoguanidine (AL0670) acetonitrile solvate, C<sub>14</sub>H<sub>18</sub>N<sub>6</sub>.CH<sub>3</sub>CN, has been determined by X-ray diffraction.

## Comment

The title compound, AL0670 acetonitrile solvate, (I), was synthesized and selected as a potent antihypertensive agent. It has a different pharmacological profile from pinacidil, although both are regarded as potassium-channel openers. The synthesis of the compound and the absolute configuration of its hydrochloride have been reported previously (Eda *et al.*, 1994).



AL0670 has more than four polymorphic forms. In this paper, the crystal structure of AL0670 acetonitrile solvate, which is one of the polymorphic forms, is reported. The molecular structure is shown in Fig. 1. The N2—C6 bond has a *cis* configuration with respect to the aminopyridyl group and the bicycloalkyl group, while it has the *trans* configuration in the case of AL0670 hydrochloride (Eda *et al.*, 1994).

The crystal structure is shown in Fig. 2. The hydrogen bond  $N5 \cdots HN6-N6$  connects symmetry-related AL0670 molecules into a ribbon along the  $a$  axis, while  $N1 \cdots HN2-N2$  does so along the  $b$  axis. Acetonitrile is bound to AL0670 via an  $N7 \cdots HN3-N3$  hydrogen bond, such that the molecular axis of acetonitrile is almost parallel to the  $a$  axis.

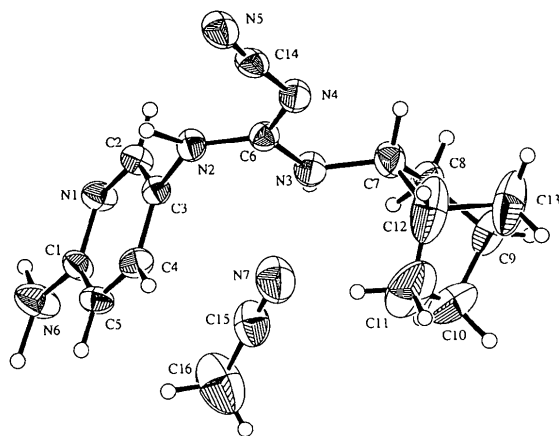


Fig. 1. A perspective view of the molecules (30% probability ellipsoids) with the atomic numbering scheme.

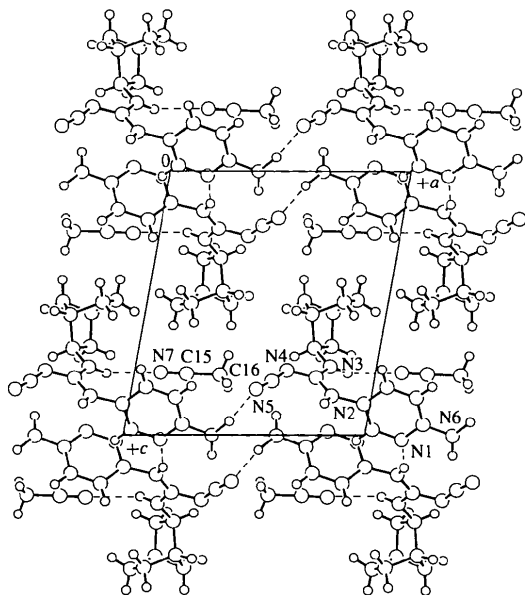


Fig. 2. The crystal structure viewed along the  $b$  axis. Hydrogen bonds are represented as dashed lines.

## Experimental

The title compound was synthesized according to the method of Eda *et al.* (1994), and recrystallized from acetonitrile.

## Crystal data

$C_{14}H_{18}N_6 \cdot CH_3CN$

$M_r = 311.39$

Monoclinic

$P2_1$

$a = 10.427(1) \text{ \AA}$

$b = 7.197(1) \text{ \AA}$

$c = 11.805(1) \text{ \AA}$

$\beta = 99.755(8)^\circ$

$V = 873.0(2) \text{ \AA}^3$

$Z = 2$

$D_x = 1.184 \text{ Mg m}^{-3}$

Cu  $K\alpha$  radiation

$\lambda = 1.5418 \text{ \AA}$

Cell parameters from 25

reflections

$\theta = 28.85\text{--}29.85^\circ$

$\mu = 0.610 \text{ mm}^{-1}$

$T = 296 \text{ K}$

Prismatic

$0.35 \times 0.30 \times 0.10 \text{ mm}$

Colorless

## Data collection

Rigaku AFC-7R diffractometer

$\omega$ - $2\theta$  scans [width (1.68

+  $0.30 \tan \theta$ ) $^\circ$ ; speed

$16^\circ \text{ min}^{-1}$  in  $\omega$ ]

Absorption correction:

$\psi$  scan

$T_{\min} = 0.844$ ,  $T_{\max} =$

1.000

1494 measured reflections

1419 independent reflections

1195 observed reflections

[ $I > 3\sigma(I)$ ]

$R_{\text{int}} = 0.012$

$\theta_{\max} = 60.05^\circ$

$h = -11 \rightarrow 11$

$k = -8 \rightarrow 0$

$l = 0 \rightarrow 13$

3 standard reflections

monitored every 150

reflections

intensity decay: none

## Refinement

Refinement on  $F$

$R = 0.045$

$wR = 0.071$

$S = 1.45$

1195 reflections

208 parameters

H-atom parameters not refined

$w = 1/\sigma^2(F)$

$(\Delta/\sigma)_{\max} = 0.03$

$\Delta\rho_{\max} = 0.17 \text{ e \AA}^{-3}$

$\Delta\rho_{\min} = -0.17 \text{ e \AA}^{-3}$

Extinction correction:

secondary

Extinction coefficient:

$2.61694 \times 10^{-5}$

Atomic scattering factors

from *International Tables*

for *X-ray Crystallography*

(1974, Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

$$B_{\text{eq}} = (8\pi^2/3) \sum_i \sum_j U_{ij} a_i^* a_j \cdot \mathbf{a}_i \cdot \mathbf{a}_j$$

	$x$	$y$	$z$	$B_{\text{eq}}$
N(1)	0.6557 (3)	0.0000	0.5143 (2)	5.23 (7)
N(2)	0.3360 (3)	0.1291 (7)	0.3500 (2)	4.72 (7)
N(3)	0.3082 (3)	-0.1389 (8)	0.2415 (3)	5.21 (7)
N(4)	0.1312 (3)	0.0380 (8)	0.2532 (3)	5.67 (8)
N(5)	0.0178 (3)	0.2997 (9)	0.3250 (4)	6.7 (1)
N(6)	0.8653 (3)	0.0861 (9)	0.4976 (3)	7.8 (1)
N(7)	0.3905 (5)	0.225 (1)	0.7375 (5)	9.6 (2)
C(1)	0.7352 (3)	0.0918 (8)	0.4555 (3)	4.87 (8)
C(2)	0.5285 (3)	0.0126 (8)	0.4752 (3)	4.96 (8)
C(3)	0.4743 (3)	0.1085 (7)	0.3795 (3)	4.14 (7)
C(4)	0.5582 (3)	0.1995 (8)	0.3171 (3)	4.73 (8)
C(5)	0.6887 (4)	0.1924 (8)	0.3558 (3)	4.99 (8)
C(6)	0.2578 (3)	0.0106 (8)	0.2817 (3)	4.48 (8)
C(7)	0.2286 (4)	-0.2753 (9)	0.1712 (4)	6.3 (1)
C(8)	0.2995 (6)	-0.463 (1)	0.1645 (4)	7.1 (1)
C(9)	0.2784 (6)	-0.503 (1)	0.0379 (5)	7.8 (1)
C(10)	0.370 (1)	-0.382 (2)	-0.0105 (6)	12.2 (3)
C(11)	0.313 (2)	-0.191 (2)	-0.0050 (6)	15.6 (4)
C(12)	0.1914 (8)	-0.226 (1)	0.0433 (7)	10.9 (2)
C(13)	0.1511 (7)	-0.421 (2)	-0.0058 (7)	12.1 (2)
C(14)	0.0768 (3)	0.1798 (9)	0.2952 (4)	5.23 (9)
C(15)	0.2836 (5)	0.220 (1)	0.7304 (5)	7.5 (1)
C(16)	0.1438 (6)	0.221 (2)	0.730 (1)	12.9 (3)

Table 2. Selected geometric parameters (Å, °)

N(1)—C(1)	1.342 (5)	N(1)—C(2)	1.332 (5)
N(2)—C(3)	1.433 (4)	N(2)—C(6)	1.349 (5)
N(3)—C(6)	1.320 (5)	N(3)—C(7)	1.452 (6)
N(4)—C(6)	1.320 (5)	N(4)—C(14)	1.306 (6)
N(5)—C(14)	1.149 (6)	N(6)—C(1)	1.364 (5)
N(7)—C(15)	1.104 (6)	C(1)—C(5)	1.397 (6)
C(2)—C(3)	1.362 (5)	C(3)—C(4)	1.398 (5)
C(4)—C(5)	1.361 (5)	C(7)—C(8)	1.547 (7)
C(7)—C(12)	1.54 (1)	C(8)—C(9)	1.502 (7)
C(9)—C(10)	1.48 (1)	C(9)—C(13)	1.47 (1)
C(10)—C(11)	1.50 (2)	C(11)—C(12)	1.49 (1)
C(12)—C(13)	1.55 (1)	C(15)—C(16)	1.456 (8)
C(1)—N(1)—C(2)	117.0 (3)	C(3)—N(2)—C(6)	124.2 (3)
C(6)—N(3)—C(7)	122.2 (3)	C(6)—N(4)—C(14)	120.0 (3)
N(1)—C(1)—N(6)	117.3 (3)	N(1)—C(1)—C(5)	122.3 (3)
N(6)—C(1)—C(5)	120.3 (3)	N(1)—C(2)—C(3)	124.8 (3)
N(2)—C(3)—C(2)	120.9 (3)	N(2)—C(3)—C(4)	121.1 (3)
C(2)—C(3)—C(4)	117.8 (3)	C(3)—C(4)—C(5)	119.0 (3)
C(1)—C(5)—C(4)	119.1 (3)	N(2)—C(6)—N(3)	119.6 (3)
N(2)—C(6)—N(4)	122.3 (4)	N(3)—C(6)—N(4)	118.1 (3)
N(3)—C(7)—C(8)	112.8 (4)	N(3)—C(7)—C(12)	115.5 (5)
C(8)—C(7)—C(12)	101.2 (4)	C(7)—C(8)—C(9)	103.3 (5)
C(8)—C(9)—C(10)	106.0 (5)	C(8)—C(9)—C(13)	104.1 (5)
C(10)—C(9)—C(13)	103.3 (8)	C(9)—C(10)—C(11)	103.5 (8)
C(10)—C(11)—C(12)	103.5 (8)	C(7)—C(12)—C(11)	108.9 (5)
C(7)—C(12)—C(13)	99.6 (7)	C(11)—C(12)—C(13)	101.6 (8)
C(9)—C(13)—C(12)	93.7 (5)	N(4)—C(14)—N(5)	173.2 (4)
N(7)—C(15)—C(16)	175.7 (8)		
N(1)—C(1)—C(5)—C(4)	0.6 (6)		
N(1)—C(2)—C(3)—C(4)	0.5 (6)		
N(2)—C(6)—N(3)—C(7)	177.6 (4)		
N(3)—C(6)—N(2)—C(3)	2.6 (5)		
N(3)—C(7)—C(8)—C(9)	-130.7 (5)		
N(3)—C(7)—C(12)—C(13)	162.3 (5)		
N(4)—C(6)—N(3)—C(7)	-2.2 (6)		
N(6)—C(1)—N(1)—C(2)	177.3 (4)		
C(1)—N(1)—C(2)—C(3)	1.2 (6)		
C(2)—N(1)—C(1)—C(5)	-1.8 (6)		
C(2)—C(3)—C(4)—C(5)	-1.6 (6)		
C(6)—N(3)—C(7)—C(8)	-162.9 (4)		
C(7)—C(8)—C(9)—C(10)	77.0 (7)		
C(7)—C(12)—C(11)—C(10)	72.3 (8)		
C(8)—C(7)—C(12)—C(11)	-65.8 (8)		
C(8)—C(9)—C(10)—C(11)	-71.7 (8)		
C(9)—C(8)—C(7)—C(12)	-6.7 (6)		
C(9)—C(13)—C(12)—C(11)	53.2 (8)		
C(10)—C(11)—C(12)—C(13)	-32.2 (8)		
N(1)—C(2)—C(3)—N(2)	-174.0 (4)		
N(2)—C(3)—C(4)—C(5)	172.8 (4)		
N(2)—C(6)—N(4)—C(14)	-2.8 (6)		
N(3)—C(6)—N(4)—C(14)	177.0 (4)		
N(3)—C(7)—C(12)—C(11)	56.4 (8)		
N(4)—C(6)—N(2)—C(3)	-177.6 (3)		
N(5)—C(14)—N(4)—C(6)	173 (3)		
N(6)—C(1)—C(5)—C(4)	-178.4 (4)		
C(1)—C(5)—C(4)—C(3)	1.1 (6)		
C(2)—C(3)—N(2)—C(6)	-90.9 (4)		
C(4)—C(3)—N(2)—C(6)	94.8 (5)		
C(6)—N(3)—C(7)—C(12)	81.4 (6)		
C(7)—C(8)—C(9)—C(13)	-31.6 (7)		
C(7)—C(12)—C(13)—C(9)	-58.5 (7)		
C(8)—C(7)—C(12)—C(13)	40.1 (7)		
C(8)—C(9)—C(13)—C(12)	55.4 (8)		
C(9)—C(10)—C(11)—C(12)	-1.7 (9)		
C(10)—C(9)—C(13)—C(12)	-55.1 (8)		
C(11)—C(10)—C(9)—C(13)	37.4 (7)		

The structure was solved by direct methods using *SHELXS86* (Sheldrick, 1985) and expanded by Fourier techniques using *DIRDIF92* (Beurskens *et al.*, 1992). The structure was refined by a full-matrix least-squares procedure. All calculations were performed using *TEXSAN* (Molecular Structure Corporation, 1992).

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: OH1073). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

## References

- Beurskens, P. T., Admiraal, G., Beurskens, G., Bosman, W. P., Garcia-Granda, S., Gould, R. O., Smits, J. M. M. & Smykalla, C. (1992). *DIRDIF92. The DIRDIF Program System*. Technical Report. Crystallography Laboratory, Univ. of Nijmegen, The Netherlands.
- Eda, M., Takemoto, T., Ono, S., Okada, T., Kosaka, K., Gohda, M., Matzno, S., Nakamura, N. & Fakaya, C. (1994). *J. Med. Chem.* **37**, 1983–1990.
- Molecular Structure Corporation (1992). *TEXSAN. Crystal Structure Analysis Package*. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- Sheldrick, G. M. (1985). *SHELXS86. Program for the Solution of Crystal Structures*. Univ. of Göttingen, Germany.

*Acta Cryst.* (1995). **C51**, 2333–2335

## Bis(5-chloro-2-nitrophenyl) Disulfide

J. GABRIEL GARCIA,\* † SIMON N. HAYDAR AND  
A. PAUL KRAPCHO

*Department of Chemistry, The University of Vermont,  
Burlington, Vermont 05405-0125, USA*

FRANK R. FRONCZEK

*Department of Chemistry, Louisiana State University,  
Baton Rouge, LA 70803-1804, USA*

(Received 6 April 1995; accepted 19 June 1995)

## Abstract

The title molecule,  $C_{12}H_6Cl_2N_2O_4S_2$ , lies on a twofold axis in the crystal. The C—S—S—C group has a skewed non-planar conformation with a torsion-angle magnitude of  $86.7(1)^\circ$ . The S—S—C bond angle is  $104.11(5)^\circ$ , and the S—S, S—C, C—N and C—Cl bond lengths are 2.0432(5), 1.789(1), 1.462(2) and 1.726(1) Å, respectively. The nitro group is rotated  $15.2(2)^\circ$  out of the plane of the phenyl ring.

## Comment

The title compound, (I), consists of two identical monomeric units linked by a disulfide bridge lying on a twofold axis. The torsion angle about the S—S bond is  $86.7(1)^\circ$ .

† Current address: Department of Chemistry, Clark Atlanta University, Atlanta, GA 30314, USA.